Foundations of Computational Materials Modelling Narasimhan Swaminathan Department of Mechanical Engineering Indian Institute of Technology Madras LAMMPS exercises 4

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5	Other potentials Domic crystals Simple pair potentials work. The electron transfer in ionic crystals is so high that the principal cohesion is from the Coulomb interactions between resulting charged ions. Alkali or alkali-earth with Group-6, Group-7. $\mathcal{V} = \frac{1}{2} \sum_{\alpha,\beta}^{N} \frac{Z^{\alpha} Z^{\beta} e^{2}}{r^{\alpha\beta}} + \frac{1}{2} \sum_{\alpha,\beta}^{nh} A_{\alpha\beta} \left(1 + \frac{Z^{\alpha}}{z^{\alpha}} + \frac{Z^{\beta}}{z^{\beta}}\right) \exp\left(\frac{\sigma^{\alpha} + \sigma^{\beta} - r^{\alpha\beta}}{\rho^{\alpha\beta}}\right) (38)$ Z ^{<i>a</i>} is the atomic number, z^{α} is number of electrons in the outermost shell, <i>e</i> is the electronic charge. σ, A and ρ are fitting parameters and depend on the ionic species.	Elst PDF Comment Comment Comparise Pages Comparise Pa
	" " from a given atom. So calculating the first term is far from trivial. • The other term is similar to the Lennard-Jones and is only short ranged Naradinhas Seminathan (IITM) ME7244 October 28, 2019 38/44	Convert and edit PORS with Acrobal PORS

So, let us continue with this, these some of these exercises. So today, last class we were just looking at this potentials called as Stillinger-Weber potential. And, and we saw another potential for the ionic crystals. So, now I want to show you a simple example where Stillinger-Weber potential is used for silicon in order to say, not do anything spectacular, but simply to print the variation of the energy with for various elastic, for various Lattice parameters. So, when you plot that you should get a U-shaped curve and the value of the Lattice constant at which the energy is a minimum is basically the Lattice constant that you get for that particular material, for that potential, yeah that is the goal.

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So, before I do that, you all should be familiar with the, the folder where LAMMPS is actually installed. So, I have installed it in some location in my computer, I have installed it right here. And if you look at it, you have the examples folder and then the potential folders. So, if you want to find out the interatomic potential for, for a specific material then you might want to start off by looking at this folder right here, a large number of potential files. And if you open one of them, say for example, this is what we are going to use, if you open them then in each of these files, you will have, you know from which article this particular potential parameters was actually taken.

So, it mean this case it was physical review B volume 31 in 1985 paper was used in order to give you the various parameters for the Stillinger-Weber potential. So, the Stillinger-Weber potential is a complicated potential with about 7 or 8 undetermined parameters which you have to give for this specific material in order for it to work. So, what we are going to do is I am just going to try to, there are several things that this exercise will do. The first thing is that it is going to, so far you have only being use the interatomic the input file. And in the input file only you described all the interatomic potential parameters also.

But now, when it comes to the potential such as the Stillinger-Weber potential, you will have to specify the potential parameters in an external file and have LAMMPS actually read the parameters from that file. A couple of things you need to keep in mind, when you are actually doing this sort of a thing, especially in the pair underscore coefficient command. The second thing is, suppose we want to do something for over a loop, over several different values of say the Lattice constant, you want it around the same file for several different Lattice constant. And then print out the energy, the Lattice constant versus the energy and make a plot of it. So, how do we do that?

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So, this is going to be exercise 13, sorry exercise 12. So, I am calling my input file as SWsi dot in. So, a lot of the stuff is pretty straightforward, so the units is metal units, atom style is atomic and the periodic boundary conditions in all directions. And then what I want to do is I want to create a Lattice with some Lattice constant a. And these, these are the basis atoms for this particular material, so it is basically a diamond structure. And then define a region and then create a simulation box, create the atoms and keep calculating, minimizing, minimize the energy for that particular value of a.

And then print out the corresponding Lattice constant and the corresponding energy that I get and do this for a large number of Lattice constants. So, I defined a variable a which is of type index, which ranges from say, 5.1 through 5.85. So, all that you have to do is in the Lattice command, you say custom and instead of giving the actual value of the Lattice parameter, you just give the variable name, so just give the variable name. So, first iteration, it will have the value 5.1, everything is a same. And then I define the basis, there is only one particular type of atom here, so basis of every all the 8 atoms is just 1. The mass of the atom is given here. And then here I define the pair style, the pair style is of type sw.

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So, as you can see from the LAMMPS website, this is the LAMMPS website showing the expression for the energy for this 3-body Stillinger-Weber potential. And you basically need to give the name of the file which contains that tabulated values of all the constants that you have to give for this particular potential. So so, you have these elements. So, Stillinger-Weber files in the potentials directory are there, so this is this is just telling you that the potentials directory of LAMMPS has Stillinger-Weber potential and you can identify it with a dot sw suffix. But then in the actual file what is there? So, you have...

So you have, see the 3-body potential, so you need to give 3 different elements. So, depending upon what you are giving here, which order you are giving here, the parameters are going to be chosen differently from this set of values. We are doing it for a homogenous material since material with only one element type. So, for this particular case, we do not

really have to worry about it. But suppose you are doing it for silicon carbide then you will have to be extremely careful. So, that is actually clearly specified here in the manual.

So, for example, if you are doing just Si Si, Si is the only thing then the first element is basically the, in the 3-body term, the first element you give is basically the one that is right in the centre. Element 2 and element 3, it could, again it is silicon-silicon in our case, because we are just simulating pure silicon then the corresponding values of epsilon, sigma, a, lambda, gamma, etcetera, etcetera. So, some of these parameters such as A, B, p and q are parameters which are going to be used only for the 2-body interactions, say, if you look at it there are so many different parameters that come here.

So, for the 2-body interactions you have A, B, p and q are the only thing that is useful parameters that are used for these 2-body interactions. Lambda cos theta naught parameters are used only for the 3-body interactions whereas the others like epsilon, sigma parameters are used for both 3-body and 2-body interactions. The most important thing is this sentence right here, so while it is specified in a very concise manner, you really have to think about it when you are actually setting up your interatomic potential file. So, for a single element simulation, you need to give only Si Si Si, it is not very complicated.

However, if your simulation contains 2 elements, say for example, silicon carbide then your every sentence in your interatomic potential file must contain an, must contain an entry and there must be 2 to the power 3 entries for each possibilities of the 3 different elements that you can have at the centre and 2 at the sides. So, you should have combinations for Si Si Si Si Si Si C and so on. So, we will, I will show you an input file which contains the for silicon carbide, just to illustrate how you do that. But for now, I just want to talk about the file which contains only silicon.

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So, once you give these parameters right here then the pair coefficients, the pair the pair style basically specifies what is the name of the interatomic potential that we are going to use. Whereas a pair coefficient, it tells several things, it tells you what type of atoms are actually interacting, what is the file from which you can obtain these potential parameters and an elemental list which basically maps the element in the interatomic potential file to the atom type that you have chosen. So, in our case, silicon is the only atom type, so there is only one atom type present in our system. So, this Si is mapped this Si is mapped to atom type 1 in our in our simulation. Suppose we did this then, then what am I doing? I minimize the structure for this particular value of a. And then I define a new variable, variable latcon equals dollar a, basically Lattice constant is equal to the value of the a.

And then in thermo style, I print the variable latcon and the corresponding potential error, so that my thermo, so that my thermodynamic file will have their information. But when I use the thermo style, I have to do a run. If I did not do a run, nothing gets printed. So, I just say run 1, that means in once, it is just going to do 1 step and it is going to print out the corresponding Lattice constant and the potential energy that it calculated at the end of the minimization process. Then I say, next a, that means of all the a's present here, it takes the next value of a and then I say clear and then I say jump SWsi dot in which means rerun the contents of this file. SWsi dot in is actually the name of this script, this script. So, it will rerun the contents of this script, but before doing that, you need to make sure that you have performed a clear. If you do not book clear, it will do complain.

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So, let us see what it does. So, let me just run it, what happens? So, it is done. So, it is giving you, its, its right for all those values of a here and if you open your logged or LAMMPS file, it is printing these potential energy for each of those Lattice parameters, which it is just of course, changing. And in all the cases, I think the minimization has just exited with energy tolerance. The energy tolerance is reached. So, it has done a good job. So now, from this you can basically extract only those lines which have the potential energy using various grep commands or any UNITS command. So, what you can do is grep, say minus al Pot energy logged or LAMMPS, so gives you all the line that is just after the sentence which has the string potential energy, but then I do not want this, I do not want that line either, so I again do grep minus v PotEng, so that it contains only the Lattice constant and the energies.

And then you can get rid of those dashes and basically make a plot of the energy for every Lattice parameter and you see that the Lattice parameter is, so the minimum value of the energy is obtained for some Lattice parameter between 4 and 5.4 and 5.45. So, if you check the literature, you will find out that the Lattice constant for which this potential was fitted is actually this value. So, see this this example took very few, did not take too long to run. So, all your simulations before starting on anything large, you should do such simple exercises to make sure that the values that is predicted by the interatomic potential, for which it is fitted for, somewhat reproduced by your simulations, only then we are using the interatomic potential correctly.

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Now, let us take a look what happens if you did not, so there is there are couple of things that I did there, right? So, I said clear, I said, next a. So, obviously next a is kind of clear, it just means use the next value of a. If I did not do the clear, UNITS command after simulation box is defined, so the previous run, it had defined a simulation box. When it is running again, if I did not clear the settings, it redefines the UNITS command, which is wrong. So, when you when you do clear, there are couple of things that basically is deleted from the system's memory and you are able to run a second simulation.

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So, basically what it does is something very specific. See, this command deletes all atoms, restores all settings to their default values and frees all the memory allocated by LAMMPS. Once a clear command is executed, it is almost as if LAMMPS were starting over again with

only the exceptions not addressed. So, there are exceptions, which means the variable and all is not deleted. It still remembers what the next a is. This command enables multiple jobs to be run sequentially from one input script. These settings are not affected by the clear command, the working directory shell command, log file status, echo status and input script variables which is basically the variable command. So, this is extremely useful. Once you know how to do stuff in a loop, you can run your simulations for various temperatures, for various pressure.

So, that is one thing that I wanted to demonstrate here. So, this is the example with this Stilling-Weber potential and the pair coefficients in this particular case just had an wild character basically saying that you have to span all possible atom types for which I am giving you the interatomic potential file here. But it is possible that your interatomic potential might require interaction between atoms of different types, for example, if you were going to simulate the interactions of ideal gas containing argon and krypton, so Kr atom type 1 and atom type, 2 then you would have to have interactions between argon argon, krypton krypton and argon krypton. And then they have different values, they will have different values.

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So, for each and every possible pairs, you will have to specify the corresponding pair coefficient the atom types there. So, so here, each star means all atom types, between all atom types. But star n means all atom types from 1 through n, including n and 1, n star means all atoms types from small n to capital N, where N is a (total) capital N is total number of atom types. So, n star N means all atom types from m to n. Sorry this should be m to n. So, so you can give here something like this.

So, there are 4 atom types, interactions between atoms of type 1 to 2 with interactions of atoms of type 3 to 4 are going to have some set of coefficients here. It is not applicable for this example, but imagine your Lennard-Jones case where you actually specify epsilon and sigma and the corresponding cut-off. So, instead of choosing each and every one, suppose you had 20 different atom types, then you could in principle do something 1 to 10, 11 to 20 to describe the interactions between 1 through 10, all atom types are 1 through 10 with all atom types of spanning from 11 through 20. Of course, you will have to give other interactions as well, this is just one set.

Student: It should be 10 power 10, 10 power 10.

Professor: But, 10 power 10 interactions, possible this is just an example. So, let us give one, seems huge. So, 1 to 2, 2 to 4, 3 to 4.

Student: So, if I...

Professor: Yes.

Student: If I am making this command...

Professor: No that is what to give those many, the idea is to give so many interactions, you do not have to like type each and every pair is what, this command helps you do. If they are having the same set of sigma epsilon values.

Student: Oh, so this is only for one particular sigma.

Professor: Yeah, yeah, exactly.

Student: So, they are changing every line?

Professor: You have to give a additional lines with the corresponding pair coefficients. You have to add other pair coefficients to take care of other interactions.

Student: When you repeat from a file that, so will it match on its own?

Professor: It will not match on its own. You have to give what element you are matching to. So, for example in this case the Si that you are giving here, the Si that you are giving here is actually the same label that appears in the potential file here, this label that is appearing here. So, that could be a and this could be a a a. Because it appears first, that element Si or a a is going to be mapped to atom type 1.

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So, what this means is one, one more way of explaining this is to actually, say it for example, just for the heck of it, 8, it works. So so, there are 8 atom types, I just set there are 8 atom

types, did you see that? I just set, there 8 atom types, basis atom 1 is of type 1, basis atom 2 is of type 2, basis atom 3 is of type 3, I I just set that and then I just set my in my pair coefficient, this is Si dot sw is the name of the file and each of these labels Si should map to the corresponding type that it is, that is generated. So, each of this is a, so you can imagine if it is more than 1.

So, if it is silicon carbide then you would, if in your input file, you you modelled Si to be of type 1 and C to be of type 2 then you would do this, but you can also do this provided you keep track of which basis is carbon and which basis is basically silica. Now, this is easier set than dump, it can get really, really confusing when you are actually doing it for these multibody potential. So, I will try to give you an example where we have, we do a test of potential where exactly the same type of input file actually has to be given.

Student: Sir if...

Professor: Yeah?

Student: If I use 1 star 2 and 3 star 4, does it mean that...

Professor: What is it, what is it? If I use?

Student: Pair coefficient, I use 1star 2 and 3 star 4.

Professor: 1 star 2 and?

Student: 3 star 4, does it mean it includes only the interaction between 1 and 2 and 3 and 4?

Professor: No. 1 2 and 3 4. All atoms of types 1 through 2 including 1 and 2 with all atoms of type 3 to 4 including 3 and 4.

Student: So, it includes 1 3, 1 4, 2 3?

Professor: 2 3, 2 4, yes.

Student: So, it does not include 1 with 2?

Professor: Yes, you would have to give that, yes.

Student: Sir, for, for this particular, can write this code? We should know what, what the structure of that file is. It is due to the same label.

Professor: Label, yes, yes. We are generating that label, you are, actually you are generating that file.

Student: But in this case, we took it...

Professor: We took it for, otherwise you will have to generate. You could call it A,B,C and put A,B,C here, it should be it should work fine.

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Now, another important form of interatomic potential is basically the embedded atom method. So, the problem is, see now all of these examples that I am giving you, I am just going to do how to specify the interatomic potential file and do a simple, plot the Lattice constant versus energy, because minimization and doing other exercises on that is kind of similar to what we have done for argon. At least the kind of fixes that you must use, all those things remain pretty much the same.

So, why do you need, there are other kinds of, right now whatever we talked about were called potentials, pair potentials which basically, a pair potentials or 3-body potentials which basically was a reflection of the fact that we could truncate our interatomic potential at some point. So, this is 2-body, this is the 3-body and then you have the 4-body and so on. We just said that anything after this is actually not important and we stop with it. So, we did Lennard-Jones 2-body and then Stilling-Weber's 3-body potential. But it so happens that for some in some cases like metals, these the higher body potentials do not converge quickly to a value.

So, you really cannot neglect all the higher, higher order potentials so to speak. But it is also inconvenient to use all the higher order potentials for obvious reasons, you have more combinations of r alpha beta to actually take in to account. And it is very difficult to come up with functional forms that will actually do this. So, people have come up with something referred to as functionals. The name functional is used wherever the interatomic potential form does not appear like what it does, here it is called a cluster potential, does not appear like this. The functional form does not appear like this then it is called a functional.

The interatomic potential form of the function does not look like that, so it is called as a functional. And you can, there are interatomic models which are actually not looking like that and they are classified into 2 categories, one is pair functionals and other is cluster functionals. So, pair functionals, as you can imagine require only, they still require only distance information, but the functional form will not look like that and cluster functional involves group of 3 or more atoms. So, an example of the embedded atom potential is like this, which is very commonly used for metals.

Embedded atoms, so it is kind of capturing the fact that your atom is actually embedded in a sea of electrons. So, basically the energy depends upon the energy of the electron gas that is basically surrounding the atom. So, that is the basic idea, of course there is a lot of physics as to why that form of interatomic potential is appropriate for such systems, we are not going to

deal, deal with that. We just want to know, what this embedded atom does and approximately somewhat understand how to use it when you are trying to do simulations with LAMMPS. So, embedded atom method, this is the functional form, it consists of a pair function plus the embedding energy term, which depends on rho alpha which is again the local density of the electron gas.

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So so, v alpha is viewed as a Coulomb interaction between atom but it is not actually z alpha, z beta and all that, something else. And U alpha is the electronic energy associated with the embedding of the atom alpha in a sea of electron or electron gas of density rho alpha, so that is what this is.

(Refer Slide Time: 28:40)

2	13 26.982 3.9860 fcc	
3	500 9.999999999999999829e-05 500 1.5000000000013e-02 6.0000000000000000e+00	
4	-1.810000000000165e+00 -1.9022962514311104e+00 -2.0085105389246820e+00 -2.1188224052313984e+00 -2.231500	9761
5	921090000 -2.34577579236836900+00 -2.46121836384519100+00 -2.57755198289909520+00 -2.69458996974087480+00 -2.812196 9428940+00	4961
6	-2.9382692775474242e+00 -3.0487289056655271e+00 -3.1675122276861316e+00 -3.2865680267814525e+00 -3.405854	0895
7	-3.525351483123794e+00 -3.6449813962152717e+00 -3.7647673910044404e+00 -3.8846712283076386e+00 -4.004673	19074
8	-4.1247588374551185e+00 -4.2449114461591932e+00 -4.3651188682812290e+00 -4.4853696925991642e+00 -4.605653	7556
9	-4.7259619714920120e+00 -4.8462861905513819e+00 -4.9666190809607258e+00 -5.0869540288152564e+00 -5.207285	0533
10	-5.3276067347616447e+00 -5.4479141517029177e+00 -5.5682028280512839e+00 -5.6884686862398155e+00 -5.808708	19968
11	-5.9289173936401660e+00 -6.0490937398082849e+00 -6.1692342062616774e+00 -6.2893361915696460e+00 -6.409397	3134
12	-6.5294153890215512e+00 -6.6493884177934319e+00 -6.7693145663374708e+00 -6.8891921546093613e+00 -7.009019	6436
13	-7.1287956242852033e+00 -7.2485188874394387e+00 -7.3681880146703520e+00 -7.4878021700893100e+00 -7.607360	2928
14	-7.7268614901096839e+00 -7.8463049511657346e+00 -7.9656899413562314e+00 -8.0850157970193663e+00 -8.204281	9206
15	52/2/2/2/0700 -8.3234877755057718e+00 -8.4426328865148434e+00 -8.5617168265690111e+00 -8.6807392230566620e+00 -8.799699 75202-00	7496
16	-8.915981244936556e+00 -9.0374341071977256e+00 -9.1562074965801230e+00 -9.2749181281770348e+00 -9.393565	8720
17	500/300+00 -9.5121506309306483e+00 -9.6306723379700543e+00 -9.7491309553319638e+00 -9.8675264723652845e+00 -9.985858	19041
18	023420er00 -1.084128289821972e+01 -1.0222334691701576e+01 -1.0340478193558795e+01 -1.0458558899670891e+01 -1.057657	6933
19	00079900+01 -1.0694532437484668e+01 -1.0812425570402638e+01 -1.0938256508113359e+01 -1.1048825441863786e+01 -1.114573	2577
	642/000+01 -1.1283378135413557e+01 -1.1400962348393875e+01 -1.1518485462374429e+01 -1.1635947735075661e+01 -1.175334	9435



pair_style buck/long	LAMMPS and described on this page. The NIST site is main	tained by Chandle	r Becker	cbecker a	t nist.gov) wh	a is good res	source	for info o	on
/coul/ong/omp command	interatomic potentials and file formats.								
pair_style buckbd/coul/gauss/dst command									
pair_style buck6d/coul /gauss/long.command	For style earn, potential values are read from a file that is in	the DYNAMO sin	gle-eleme	ent funcfi t	ormat. If the D	DYNAMO fil	le was c	created b	by a
pair, style lýcharmm	Fortran program, it cannot have "D" values in it for expone	nts. C only recogni	zes e or	'E' for sci	entific notatio	.n.			
/coul/charmm command	Note that unlike for other potentials, cutoffs for EAM pote	entials are not set i	n the pair	style or p	air_coeff.com	mand; they	are spe	cified in	the
pair_style ljicharmm /coui/charmm/intel.command	potential files themselves.								
pair_style lj/charmm kou/kharmm/ik command	For style earn a potential file must be assigned to each U p	air of atom types b	yusing on	e or more	pair_coeff co	mmands, ea	ch with	asingle	
nair style Litharmm	argument:								
/coul/tharmm/omp command	A a filename								
pair_style lj/charmm									
courterarmine improce communia	Thus the following command								
/coulitharmm/implicit/kk									
command	pair_coeff *2 1*2 cull.eam								
pair_style ij/charmm /coui/charmm/impilicit/omp									
command	will read the cuu3 potential file and use the tabulated Cu v	alues for F, phi, rho	that it co	ntains for	type pairs 1,	l and 2,2 (ty	pe pairs	s 1,2 and	12,
pair_style lj/charmmycoul/long	ignored), see the pair_coeff doc page for alternate ways to LAMMPS be Culatoms, Different single-element files can	specify the path to be assigned to diffe	or the pool	ential hie.	in effect, this model an alig	makes atom v system. Th	types:	1 and 2 ii	in iate
nair stula Litharmen	potentials for type pairs with I != J is done automatically ti	e same way that th	ne serial D	YNAMO	code original	y did it; you	do not	need to :	spe
/coul/long/gpu command	coefficients for these type pairs.								
pair_style lj/chamm/coul/long	Freed Glassie the extentials disastees of the LANBARC dist		M.		10 circle cir		61+ 1+ 6+		
/intel command	follows:	oodon nave an .e.	in sunx	ADTINA	no sirgie ele	nent inntit i	HEP IO	rmatieu	1 92
command									
pair_style §/chamm	 line 1: comment (ignored) 								
/couliforg/opt command	 line 2: atomic number, mass, lattice constant, lattice by line 2: Nide side No do antelli 	pe (e.g. FCC)							
pair_style lj/charmm /coulifong/omp.command	 ane a: remo, orma, rer, of, CUTOTT 								1
pair style Lichamm/coulinsm	On line 2, all values but the mass are ignored by LAMMPS	The mass is in mas	s units, e,	g. mass nu	mber or gram	s/mole for n	metalur	nits. The	cut







And in LAMMPS, the interatomic potential for these systems are actually given in the form of a table, it is huge table. There are different formats in which embedded atom method, potential files can be given and this is one of the formats that LAAMPS can understand and use. So, I will open the file. So, for the style eam, a potential file must be assigned to each I, I pair of item types by using one or more pair coefficient commands with a single argument, which is basically the file name.

So, when you look at the input file, I have done exactly the same thing that I did with silicon here, where I just want to find out the energy versus Lattice constant curve. So, I have a variable index and so on and so forth, let us not. The only, the only difference is that I just construct an fcc Lattice and the only thing that I had to give here is a dollar a. Basically, the lattice parameter. And another thing that I want you to notice is that, here I had given the region in the default units, which is Lattice units, not units box, the Lattice units. Because, if it is units box then I have to every time calculate what my box size is.

When I say, Lattice units, it is just going to be automatically taking 10 times the unit cells in each of the directions. So, I give star star and then call the corresponding file which is there in the same folder called Al underscore jnp dot eam. Perform minimization and everything like what I did for this silicon file. And if you run it, you are going to get the energy variation, but it is a good idea to look at the, let us look at the, so, you get something that looks like this, when you plot the energy versus Lattice constant. The first thing that you notice is that the Lattice constant is somewhat a little bit less than 4.

If you open up the, the potential file, the first line in the potential fi is basically a comment and it is not used anywhere. The second line basically gives you atomic number, atomic mass, the Lattice constant that was obtained from this potential and the crystal structure that is used. And then you have one another line which tells you, how the function, it is a tabulated value of the function f of rho, for various values of rho and the function, another function z of r, for various values of r. It is basically the part of the 2-body potential phi.

So, if you take a look at this, it says that there are 500 entries for f of rho. So, there should be 500 entries starting from here, so going like this. So, if you go 1,2,3,4,5 and so on, until, until somewhere here. So, there will be, there will be exactly 500 entries and then another 500 entries for the, the value of the function and then the corresponding values of, let me take a look at the file, the density function rho, rho of r. The density function rho of r, how rho

varies with r, the effective charge function, how that varies with r and the embedding function f of rho, how it varies with rho.

And how rho and r are split is there in this information, d rho and d r contains that information, which is d rho, how the density is split and how the distance is actually split. So, that is the embedded atom potential and most of the time you will be able to obtain these embedded atom potentials for, for which, for whichever ones it exists.

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	NT: FUNCTIONS FOR ALUMINUM 8/26/86 FROM JACOBSEN, NORSKOV, AND PUSKA
2	13 26.982 3.9860 fcc
3	500 9.9999999999999999829e-85 500 1.50000000000013e-82 [6.000000000000000000000000000000000000
4	-1.8100000000000165e+00 -1.9022962514311104e+00 -2.0085105389246820e+00 -2.1188224052313984e+00 -2.2315009761 931890e+00
5	-2.3457767923683690e+00 -2.4612183638451910e+00 -2.5775519828990952e+00 -2.6945899697408748e+00 -2.8121964961
6	-2.9302692775474242e+00 -3.0487289056655271e+00 -3.1675122276861316e+00 -3.2865680267814525e+00 -3.4058540895
7	S10805e+80 -3.5253351483123794e+80 -3.6449813962152717e+80 -3.7647673918044484e+80 -3.8846712283076386e+80 -4.8846739074
	910533e+00
8	-4.1247588374551185e+00 -4.2449114461591932e+00 -4.3651188682812290e+00 -4.4853696925991642e+00 -4.6056537556 256671e+00
9	-4.7259619714920120e+00 -4.8462861905513819e+00 -4.9666190809607258e+00 -5.0869540288152564e+00 -5.2072850533
10	-5.327606/3347616447e+00 -5.4479141517029177e+00 -5.5682028280512839e+00 -5.6884686862398155e+00 -5.8087080068
11	-5.9289173930401660e+00 -6.0490937398082849e+00 -6.1692342062616774e+00 -6.2893361915696460e+00 -6.4093973134
12	-6.5294153890215512e+00 -6.6493884177934319e+00 -6.7693145663374708e+00 -6.8891921546093613e+00 -7.0090196436
13	-7.1287956242852033e+00 -7.2485188074394387e+00 -7.3681880146703520e+00 -7.4878021700893100e+00 -7.6073602928
14	-7.726811491096839e+00 -7.8463049511657346e+00 -7.9656899413562314e+00 -8.0850157970193663e+00 -8.2042819206
15	532/2/2+00 -8.3234877765057718e+00 -8.4426328865148434e+00 -8.5617168265690111e+00 -8.6807392238566620e+00 -8.7996997496
16	//2/398+80 -8.9155981244936556e+80 -9.0374341071977256e+80 -9.1562074965801230e+80 -9.2749181281770348e+80 -9.3935658720
17	856/360+80 -9.511506309306483e+80 -9.6306723379700543e+80 -9.7491309553319638e+80 -9.8675264723652845e+80 -9.9858589041
18	825425e+80 -1.0104128289821972e+01 -1.0222334691701576e+01 -1.0340478193558795e+01 -1.0458558899670891e+01 -1.0576576933
19	665990e+01 -1.0694532437484668e+01 -1.0812425570402638e+01 -1.0930256508113359e+01 -1.1048025441863786e+01 -1.1145732572
VILL	64270e##01 -1.12833781354513557e+01 -1.1400962348393875e+01 -1.15138485462374429e+01 -1.1635547735075661e+01 -1.1783349435 43555e+01
AR	np.eam" 304L, 35743C NPTEL



Usually it can be a hard time to find for arbitrary alloys and you should, what is the name of that, yeah NIST, Interatomic Potential Repository basically contains all possible interatomic potentials that might be there, unless new ones are invented and they have not yet been up updated, you will find most of the information in this website. So, you can take a look at this and get all the information of the potentials for which you have the interatomic (potential), elements for which the interatomic potentials basically existed.

Student: Number?

Professor: Yeah.

Student: In the precious cases, we did not mentioned atomic number, in this case why is it necessary?

Professor: It is probably used by the interatomic potential somewhere. I do not really know how the interatomic potential has actually been developed. So, it is possible that it is going to get used somewhere in the interatomic potential functional form, which is inside the, which is inside LAMMPS.

Student: And in case of Lennard-Jones, we were writing pair style, nj slash cut and the cut off length.

Professor: Correct, i j epsilon, sigma and then the corresponding cut-off radius, yeah.

Student: So, in case of eam, we do not need any cut off distance to...

Professor: It is there it is there actually. It is right, you do not have to explicitly specify it, it is there in the file, this is the cut-off.

Student: So we do not have to specify it in pair style?

Professor: Yeah you do not have to specify in pair style for the eam potential. So, now you can imagine, we plotted the variation of energy with Lattice constant. So, this amounts to plotting the variation of energy with volume as well, because a cube is nothing but the volume. So, which means you should be able to find out approximately the bulk modulus of the material.

If you just plotted this energy for various volumes then dou e by dou v, 1 by v times dou v by dou v should actually give you the bulk modulus of the material at low temperatures and for material, at low temperatures, for because, and for metals and ceramics, for really low temperatures, like close to 300 or 400 Kel, until, unless you are like close to its melting point for extremely, for temperatures far away from the melting point, the bulk modulus and the Young's modulus, the mechanical properties of the system are not significantly affected by temperature. So, you can get a rough estimate of these mechanical properties also by making a plot of this energy versus the volume. So, I think that that you should be able to do.