

Fundamentals of Spectroscopy
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Lecture-12

Rotational Spectroscopy: Correspondence between Linear Motion and Rotational Motion

Hello everyone, welcome to this lecture. In this lecture, we will start a new module where we will look into rotational spectroscopy in detail. So, as we have already discussed before we can have rotational spectroscopy in the microwave region of the electromagnetic spectrum.

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The slide is titled "Rotational Spectroscopy" and contains several diagrams and equations:

- Energy Level Diagram (Top Left):** Shows three energy levels labeled E_{rot} , E_{vib} , and E_{elec} . The energy differences are indicated as $\Delta E_{rot} \ll \Delta E_{vib} \ll \Delta E_{elec}$.
- Electromagnetic Spectrum Chart (Top Right):** A logarithmic scale of frequency (Hz) from 10^0 to 10^{16} and wavelength (m) from 10^0 to 10^{-16} . Regions shown include Radio waves, Microwave, IR, Visible spectrum, UV, X-rays, and Gamma rays. Handwritten notes indicate the microwave region is between 1000 MHz and $300,000 \text{ MHz}$.
- Linear Motion Diagram (Bottom Left):** Shows a mass m moving with velocity v . The equation $v = \frac{dx}{dt}$ is written.
- Angular Motion Diagram (Bottom Middle):** Shows a mass m moving in a circle of radius r with angular velocity ω . Equations include $\omega = \frac{d\theta}{dt}$, $\omega = \frac{v}{r}$, $v = \frac{\omega r}{2\pi}$, and $\omega = \frac{v}{r}$.
- Speaker (Bottom Right):** A photograph of Prof. Dr. Anirban Hazra.

So, let us look into the electromagnetic spectrum again. So, this is the microwave region. So, the frequency range for the microwave electromagnetic region is from 1000 megahertz to 300,000 megahertz and we know that 1 megahertz equals 10 to the power 6 hertz. So, roughly speaking, so, the microwave region falls between 10 to the power 9 to 10 to the power 12 hertz and in this electromagnetic spectrum the frequency is increasing towards the left and as we have also discussed before for this different transitions like electronic rotational vibrational.

So, the ΔE electronic is much better than ΔE vibrational which is much greater than ΔE rotation. So, if we draw the energy levels, let us say we have this electronic energy levels E electronic. So, this is my ΔE electronic. So, for each electronic energy level, we have these different vibrational energy levels this is E vibrational. So, a transition here the energy difference is ΔE vibrational, and for each vibrational level, there are different rotational levels.

So, this is ΔE rotational, and we are plotting here the rotational levels. So, because today we are talking about rotation or rotational spectroscopy, let us consider a rotating rigid body the classical mechanics of rotational motion of a rigid body is relatively complicated. So, it is useful to note the extensive correspondence between a linear motion and the angular motion or the rotational motion of a point particle of mass m . So, let us consider this particle of mass m is rotating in a circle of radius r .

So, for us small displacement x from any point on the circle, we can assume this small segment is linear. That is, although this is linear it is part of this circumference of the circle. So, we can write $\sin \theta = x / r$. So, because we have considered this displacement to be very small. So, the θ actually is also very small, and because it is small, we can write $\sin \theta$ is approximately equal to θ . So, because $\sin \theta$ is approximately equal to θ , we can write $\theta = x / r$. So, here r is a constant as the particle is rotating in a circle of constant radius.

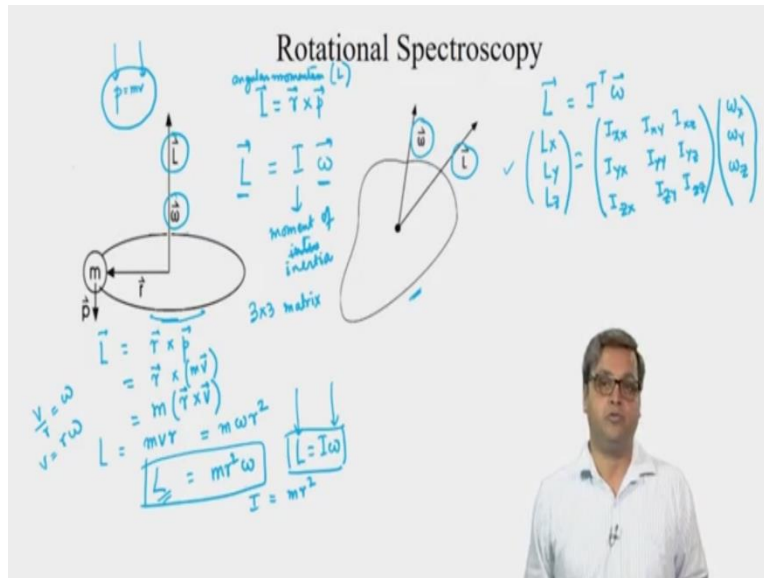
However, x is a variable and so is θ , so thus we can think the distance in linear motion is analogous to the angle swept out by the particle in angular or rotational motion, and we also know any linear motion is associated with a linear velocity v , which is given by $x \dot{\theta}$ that is dx/dt . So, for angular motion, similarly, we have a velocity, which is known as angular velocity, which is denoted by ω and ω is given by as we see from the analogy from the linear motion, ω is given by $\dot{\theta}$ that is $d\theta/dt$.

So, because $\theta = x / r$ we can write $\omega = d(x / r) / dt$ and as I already mentioned, the particle of mass m is rotating in a circle of constant radius. So r is a constant. So you can write $1 / r \cdot dx / dt$ and this dx / dt is the velocity, as a linear velocity. So we can write $\omega = v / r$. So we can think this angular velocity as the number of radians of angle swept out in any time by a

rotating system. So we can also calculate that for a particle moving with a linear velocity v , the frequency ν or how many revolutions of circumference $2\pi r$.

That is the circumference of the circle with radius r is completed per unit time. So, ν is $v / 2\pi r$, and as for every revolution, it sweeps out an angle of 2π . or in other words, we know that ω equals $2\pi\nu$. So, we can write ω equals $2\pi\nu$ and ν is $v / 2\pi r$. So, we can again show that ω equals v / r .

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So, as a linearly moving body has linear momentum given by p equals mv . Our rotating body has also momentum known as the angular momentum which is denoted by L . So, we can see, L is given by r cross p . So if r and p are in the same direction, then the cross border becomes 0. So L become 0 otherwise L exists. So for a single particle I shown here in this figure, we can see, the ω and L are vectors that point out of the plane of rotation.

However, we should remember that if an extended object is rotating this ω and L need not point in the same direction and so, in general L is related to ω with the relation L equals $I\omega$. So, here we see this angular momentum L and the angular velocity ω are vectors. So, if they are in the same direction, as we see in this figure where single particle of mass m is rotating, then I is a scalar quantity, but in general, they are not in the same direction as mentioned, for this extended object, for example, a molecule.

So, I is defined as a tensor and I is known as the moment of inertia so, this tensor is represented by a matrix. So, if you consider a 3 dimensional space, it is a 3×3 matrix. In other words, in a 3×3 matrix, we have 9 components. So we can write $L = I \omega$, which is a tensor ω is a vector in 3 dimensional space we can write L as $L_x \ L_y \ L_z$ equals I is a 3×3 matrix that is $I_{xx}, I_{xy}, I_{xz}, I_{yx}, I_{yy}, I_{yz}, I_{zx}, I_{zy}, I_{zz}$ and then the angular velocity which is also vector we can write $\omega_x \ \omega_y \ \omega_z$.

So, in a 3 dimensional space we can represent $L = I \omega$ this way. So, now let us go back to a single particle rotating in a circle. So, $L = r \times p$. So, p we know is p is given by $p = mv$ so, we can write that is $r \times mv$, we can write this as $m \ r \times v$ and because r and v are perpendicular. So we can write $L = mrv$, and as we know $v / r = \omega$ or $v = r \omega$ we can write $L = m \omega r^2$.

So, if we compare this expression that is $L = m r^2 \omega$ with the other expression that is $L = I \omega$. So, we can see the analogy that $I = m r^2$ and also, L is the angular momentum and p is the linear momentum. So we see that $p = mv$. So linear momentum and is a linear velocity. So, we have angular momentum L , we have angular velocity ω . So, comparing this $p = mv$ with $L = I \omega$, grossly we can think what a mass is for linear motion, moment of inertia is for rotational motion.

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Rotational Spectroscopy $\hbar = \frac{h}{2\pi}$

Rigid Rotor Approximation

$$L = I\omega$$

$$L = m_1 r_1^2 \omega + m_2 r_2^2 \omega$$

$$L = (m_1 r_1^2 + m_2 r_2^2) \omega$$

$$I = m_1 r_1^2 + m_2 r_2^2$$

$$I = \sum_i m_i r_i^2$$

$$E = \frac{1}{2} I \omega^2 = \frac{L^2}{2I}$$

$$E = \frac{L^2}{2I} = \frac{J(J+1)\hbar^2}{2I}$$

$$E = \frac{h^2}{8\pi^2 I} J(J+1)$$

$J = 0, 1, 2, 3, \dots$

$J=0 \rightarrow E=0$

So, now moving back to polyatomic molecules, so, diatomic molecules are the easiest examples of polyatomic molecules. So, let us start with diatomic molecules and also although the diatomic molecules undergo simultaneous rotation and vibration, let us just consider the rotational effects for now. So, this is known as the rigid rotor approximation so, we will first consider diatomic rigid rotors which means the bond length is not changing just the molecule is rotating.

So, for a diatomic molecule with masses m_1 and m_2 and bond length R and such that m_1 is r_1 distance away from the center of mass and m_2 is r_2 distance away from the center of mass. So, if this molecule is rotating, and because both the atoms are rotating with constant angular velocity ω , so we can write $L = m_1 r_1^2 \omega + m_2 r_2^2 \omega$ that is $m_1 r_1^2 \omega + m_2 r_2^2 \omega$. So, if we compare this expression with $L = I \omega$, so we can get or we can see that $I = m_1 r_1^2 + m_2 r_2^2$.

So, when we were talking about a single rotating particle, we saw I is $m r^2$, where mass where m was the mass of the particle. And when we have a diatomic molecule with mass is m_1 and m_2 then I is $m_1 r_1^2 + m_2 r_2^2$. So, in general, the expression of I for a polyatomic molecule is $I = \sum_i m_i r_i^2$, where r_i is the distance of the highest particle of mass m_i from the center of mass of the system. So after we have considered the angular momentum L let us now consider the energy of a rotating diatomic rigid rotor.

So a rotating molecule has no potential energy. So normally energy is given by the sum of kinetic energy plus potential energy. In this particular case, the potential energy is 0. So, the energy or the total energy counts only from the kinetic energy, so thus we can write $E = \frac{1}{2} m_1 v_1^2 + \frac{1}{2} m_2 v_2^2$ and putting the value of v we can have $\frac{1}{2} m_1 r_1^2 \omega^2 + \frac{1}{2} m_2 r_2^2 \omega^2$.

So, that is $\frac{1}{2} m_1 r_1^2 \omega^2 + \frac{1}{2} m_2 r_2^2 \omega^2$ and as we know, this $m_1 r_1^2 + m_2 r_2^2$ is nothing, but the moment of inertia I so the energy becomes $\frac{1}{2} I \omega^2$. So, now, let us look into this expression and the analogy from the linear motion, for linear motion the kinetic energy is given by equals half mv^2 . So, as we saw, the analogous part of mass in linear motion is moment of inertia in the rotational motion and here we have linear velocity here we have angular velocity.

So, half mv^2 in linear motion, it is half $I \omega^2$ in angular motion. So, we can write this half $I \omega^2$ as $\frac{L^2}{2I}$ that is $\frac{L^2}{2I}$. So, we know $I \omega$ is the angular momentum L . So this energy expression becomes $\frac{L^2}{2I}$, again we know in linear motion, the linear momentum p . So energy or the kinetic energy is related to $\frac{p^2}{2m}$. so here the kinetic energy in the rotational motion is related to the angular momentum by $\frac{L^2}{2I}$.

So, now, let us move on to quantum mechanics. Although a complete quantum mechanical description can be obtained by solving the Schrödinger equation for a rotating linear system. At first it is more informative to see what is obtained by applying Bohr condition that the angular momentum is quantized. We know the angular momentum is quantized in the units of \hbar that is $\frac{h}{2\pi}$. So, the quantized angular momentum condition requires L that is angular momentum is $J \hbar$, where J is the rotational quantum number and J can take values like 1 2 3.

So, we can write $L = J \hbar$ as $L = \sqrt{J(J+1)} \hbar$ but the correct quantum mechanical treatment however, gives if J is an integral quantum number, the angular momentum L is given by not as square root of $J^2 \hbar^2$, but $\sqrt{J(J+1)} \hbar$. So, we can see that energy which is given by $\frac{L^2}{2I}$ we can write as $\frac{J(J+1) \hbar^2}{2I}$.

square $h^2 / 2I$, that is $h^2 / 2I J \times J + 1$ and because h is $h / 2\pi$, so we can write this as $h^2 / 8\pi^2 I J(J + 1)$

So, we can obtain the same expression of energy that is equals $8\pi^2 I J(J + 1)$ by solving the Schrodinger equation. So, because J is the rotational quantum number we see that as the value of J increases, the energy increases. So, if we go from 0 to 1 to 2 to J equals 3, the energy of this rotating particle or the rotating a molecule will increase. So, when $J = 0$, then the energy expression becomes 0 or as if the molecule is not rotating.

So, now, if we look into the angular momentum expression, we see the angular momentum increases with increasing J . So what does it mean? Angular momentum is given by $I\omega$ and I is nothing but related to the masses of the atoms and the distance or the radius r which is constant for rotational motion. So, that will not change. So, if I change J , what will change is the angular velocity. So, what we might think is as we have increased in J or increased the value of J , the molecule will start rotating in a like a larger velocity, larger angular velocity. So, before we end this lecture.

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Rotational Spectroscopy

Correspondence between linear motion and angular motion

| Linear Motion | | Angular Motion |
|------------------------------|----------------|---|
| distance $\rightarrow x$ | Position | angle $\rightarrow \theta$ |
| $v = \dot{x} = dx/dt$ | Velocity | $\omega = \dot{\theta} = d\theta/dt \Rightarrow \omega = v/r$ |
| m | Mass | I |
| $p = mv$ | Momentum | $L = I\omega$ |
| $\frac{1}{2}mv^2$ $p^2/2m$ | Kinetic Energy | $\frac{1}{2}I\omega^2$ $L^2/2I$ |

Let us look back to the correspondence between the linear motion and the angular motion. So, if you want to define position in linear motion, it is defined in terms of distance, that is x . In angular motion, it is defined in terms of angle that is given by θ . In linear motion, there is a

linear velocity which is given by v that is \dot{x} that is $\frac{dx}{dt}$ in angular motion, there is an angular velocity ω that is given by $\dot{\theta}$ that is $\frac{d\theta}{dt}$, and we saw that $\omega = \frac{v}{r}$. The mass for a linear motion is the mass m .

The analogous thing in the angular motion is the moment of inertia. So, the momentum or the linear momentum for linear motion is $p = mv$. With this correspondence we can say the angular momentum L equals what is equivalent to mass that is I and what is equivalent to linear velocity is linear motion and angular of motion we have angular velocity So, equals how you make similarly for kinetic energy, expression is $\frac{1}{2}mv^2$, or $\frac{p^2}{2m}$ and an angular motion, we have write $\frac{1}{2}I\omega^2$, or $\frac{L^2}{2I}$.